

Contents

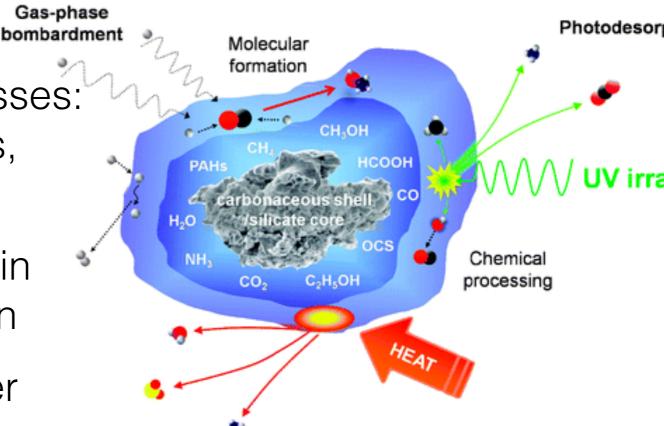
- Experimental setups
- Types of studies
- Deriving fundamental chemistry (Adi Ding)

Recap last week

 Molecules freeze out on dust grains forming ice mantles

 Ice mantles experience processes: radiation, atom bombardments, thermal processing

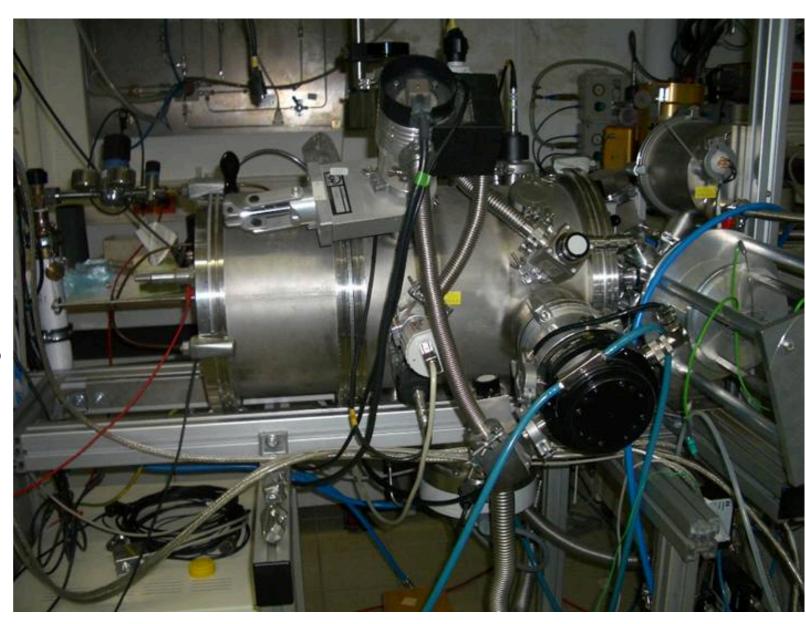
- Ices can be observed directly in infrared (vibrational) absorption
- Solid state features are broader than gas features
- Ice chemistry required to explain complex molecule abundances!



Laboratory astrochemistry

- Primary focus: solid state (ice or PAH) chemistry in a controlled environment
- Experimental work easier than mathematical/analytical derivation!
- Goals
 - Identification and quantification of vibrational features
 - Testing astrochemical processing (formation, destruction, diffusion, segregation, thermal/nonthermal desorption)
 - Deriving rate equations

 Setups big and complex: why?
what components do we need?

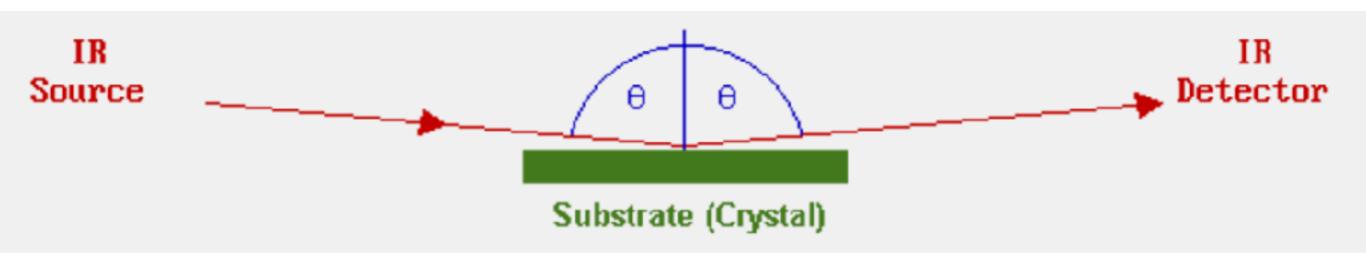


- Largest component: pumps => vacuum
 - High vacuum (~10⁻⁷ 10⁻⁸ mbar) vs ultrahigh vacuum (~10⁻¹⁰ - 10⁻¹¹ mbar): sea level ~1 bar
 - Reason: "pollution" ice by air molecules, primarily H₂O: HV experiments have to be run in a few hours and ice needs to be thick in comparison (>0.1 μm or ~3000 L, L~few monolayers)
 - UHV only possible since the last decade

- Closed-cycle cryostat: usually helium => cooling down to <15 K
- Heating device for controlled temperature rise
- Vacuum chamber with substrate: non-reactive material, e.g. gold-coat
- Outlet for connecting gas bottles

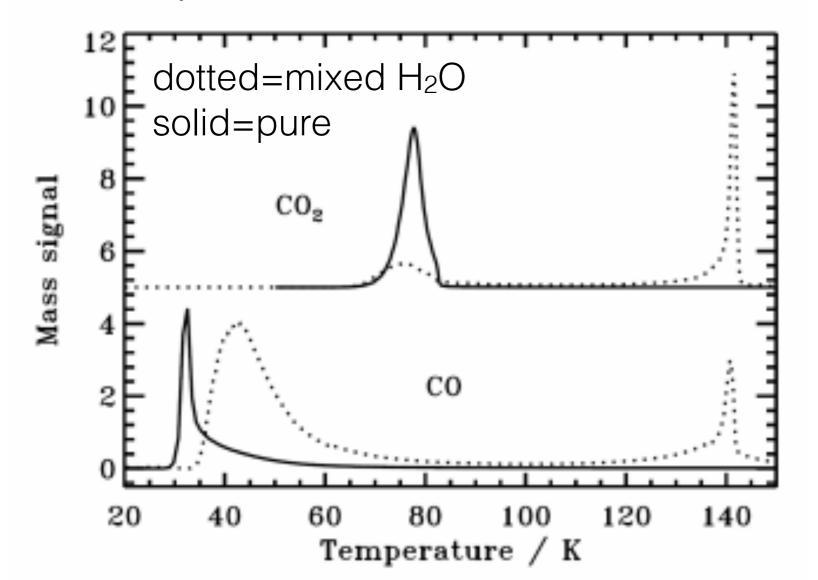
- Measuring devices:
 - FTIR spectrometer: ice-phase
 - Quadrupole Mass Spectrometer (QMS): essentially measuring ions (gas) mass/charge ratios
- Processing sources: UV lamp, atomic source

 Spectroscopy: transmission or reflection-absorption (RAIRS)



- Advantage RAIRS: increase path-length and thus sensitivity of monolayers
- Disadvantage RAIRS: no direct comparison to astrophysical spectra => more suitable for study ice processes

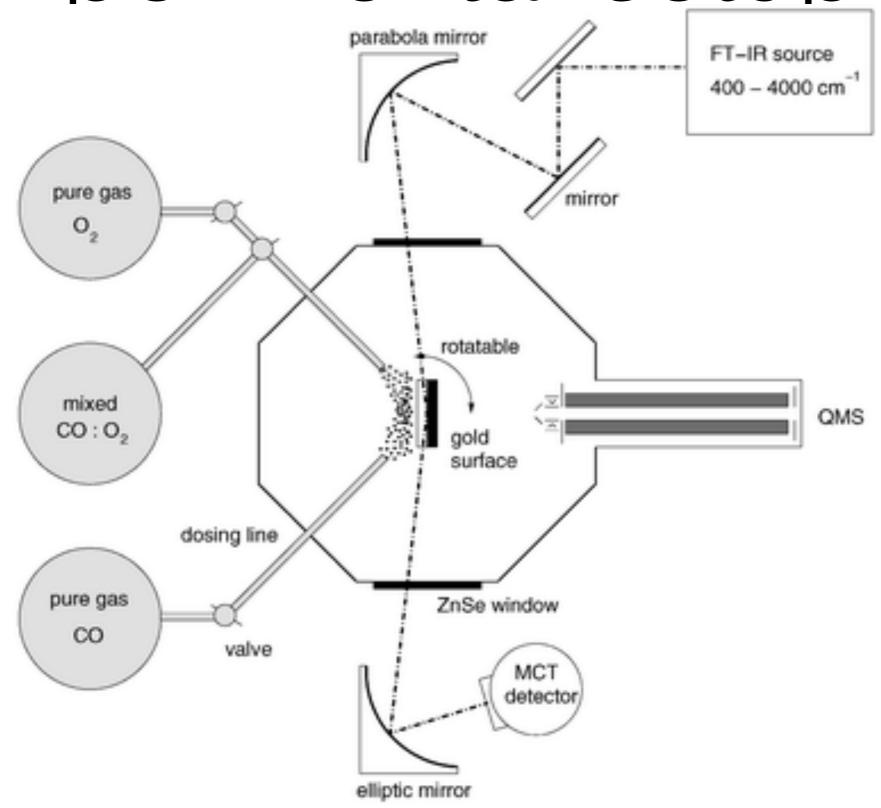
 QMS for a TPD (temperature programmed desorption) experiment: monitor gas-phase species desorbed from the ices

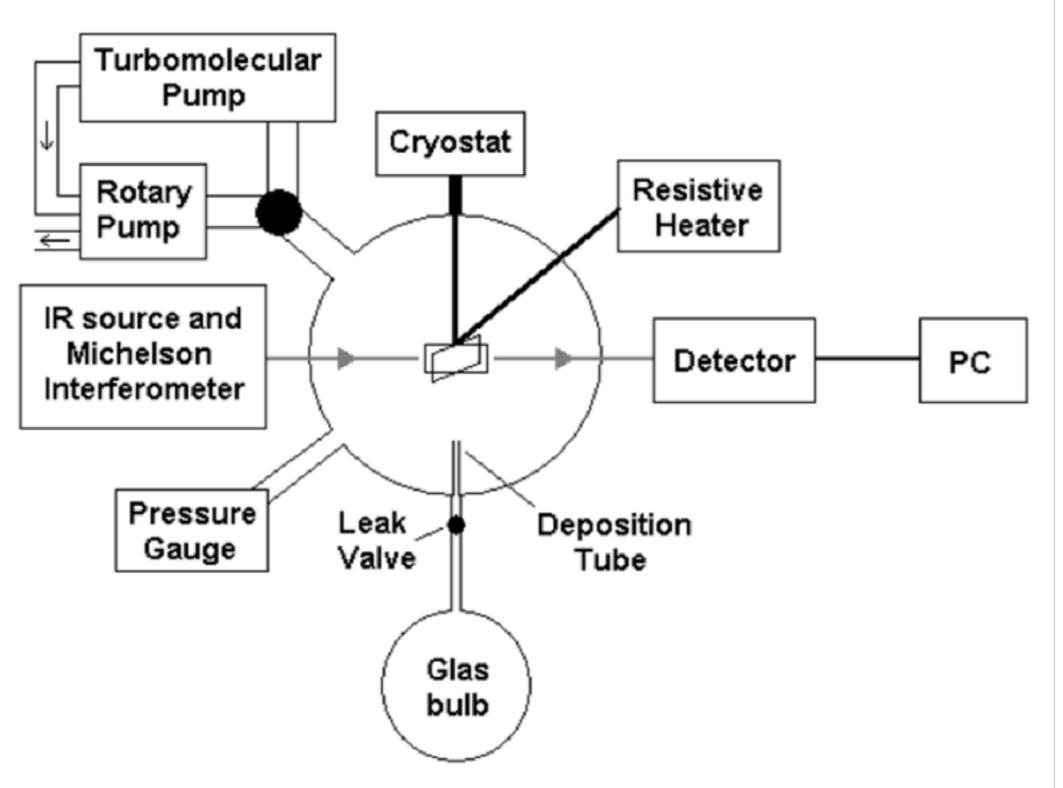


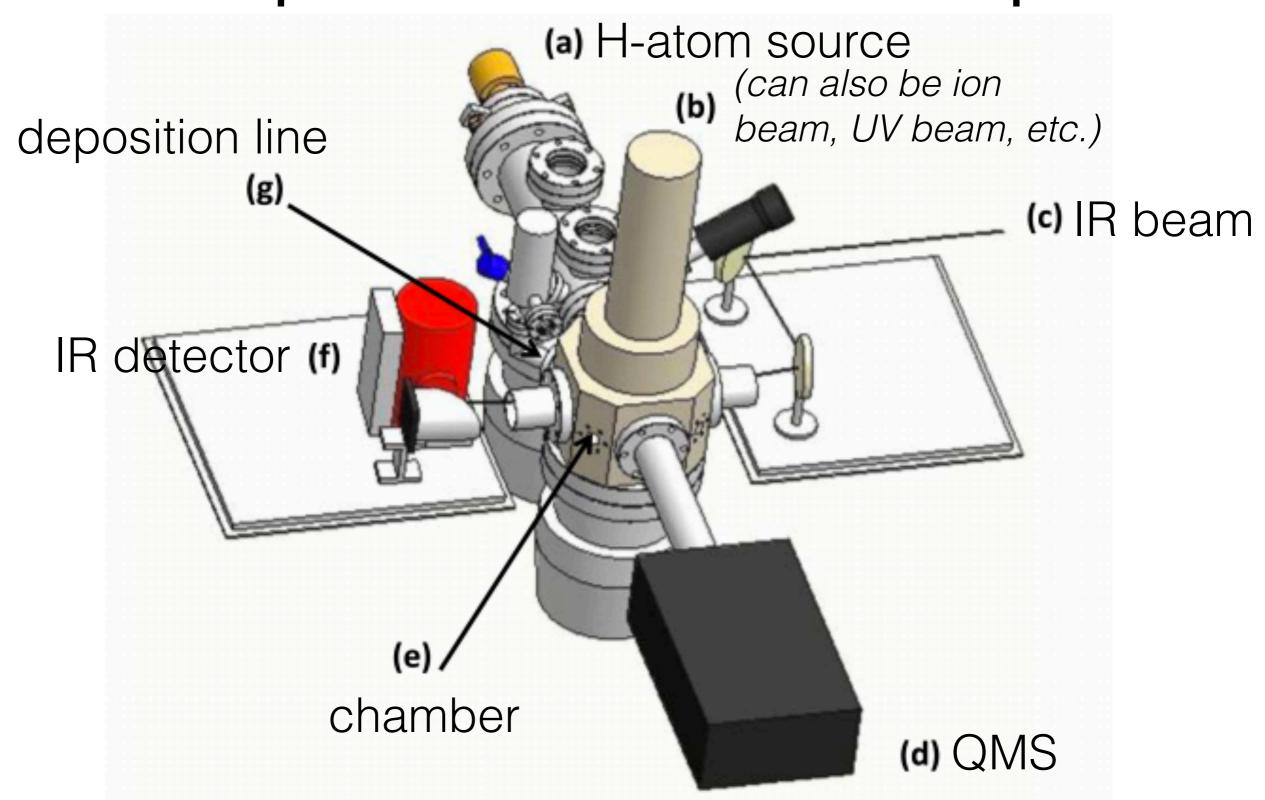
TPD vs spectrometry:

- possible reactions during TPD
- similar masses cannot be distinguished

Oberg et al. 2009





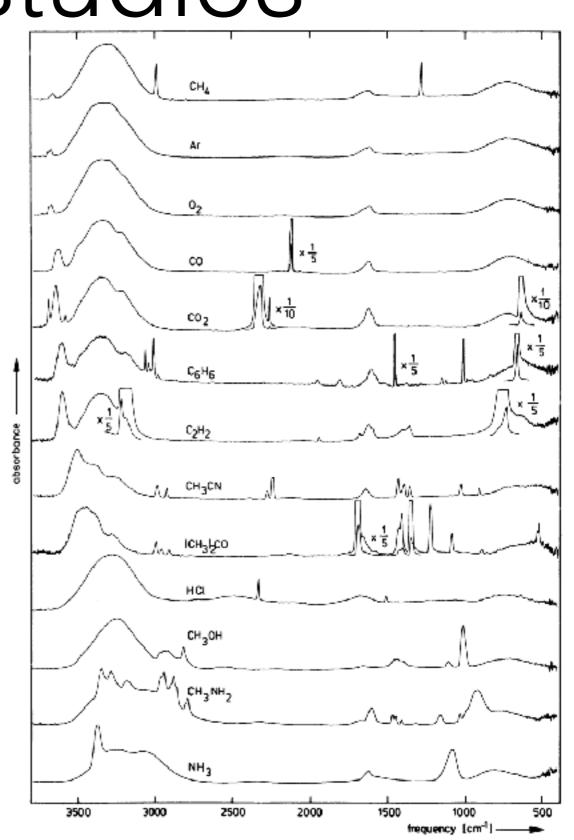


 Identification infrared features and mixing ratios

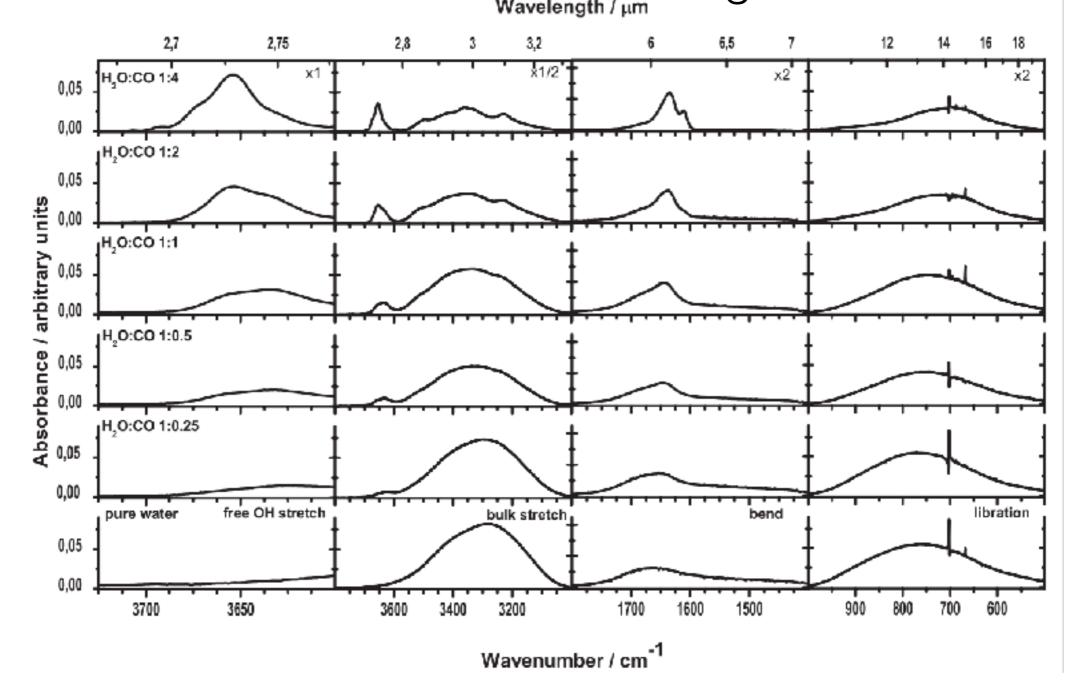
Note: absorption positive!

Procedure:
 isolate, cool down,
 vacuum, freeze-out,
 stabilize, spectrum,
 desorb

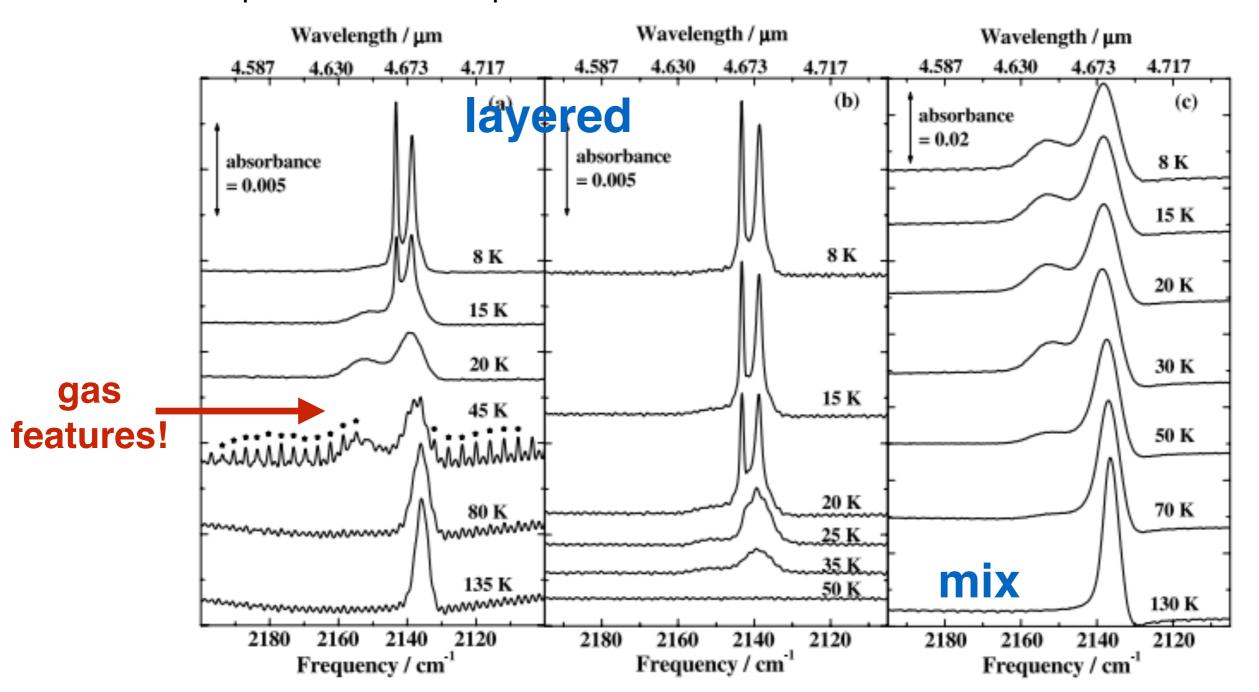
Hagen et al. 1983



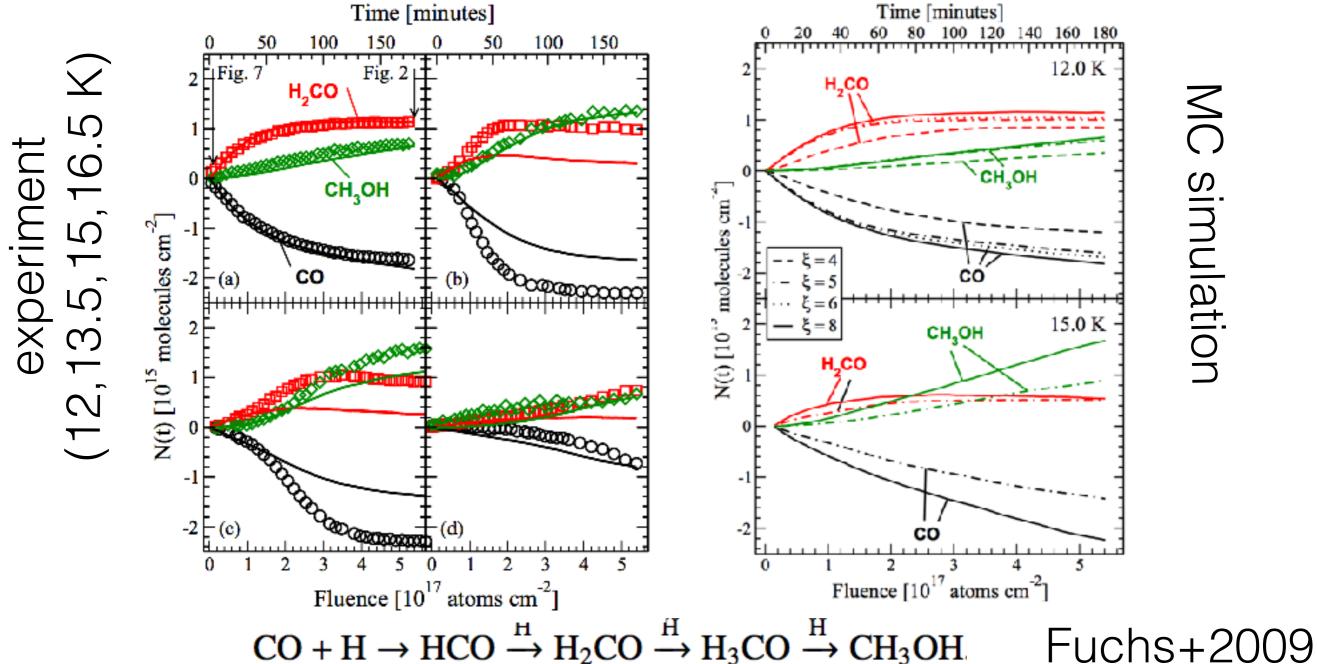
• Identification features and mixing ratios



Temperature dependence ice features of CO/H₂O

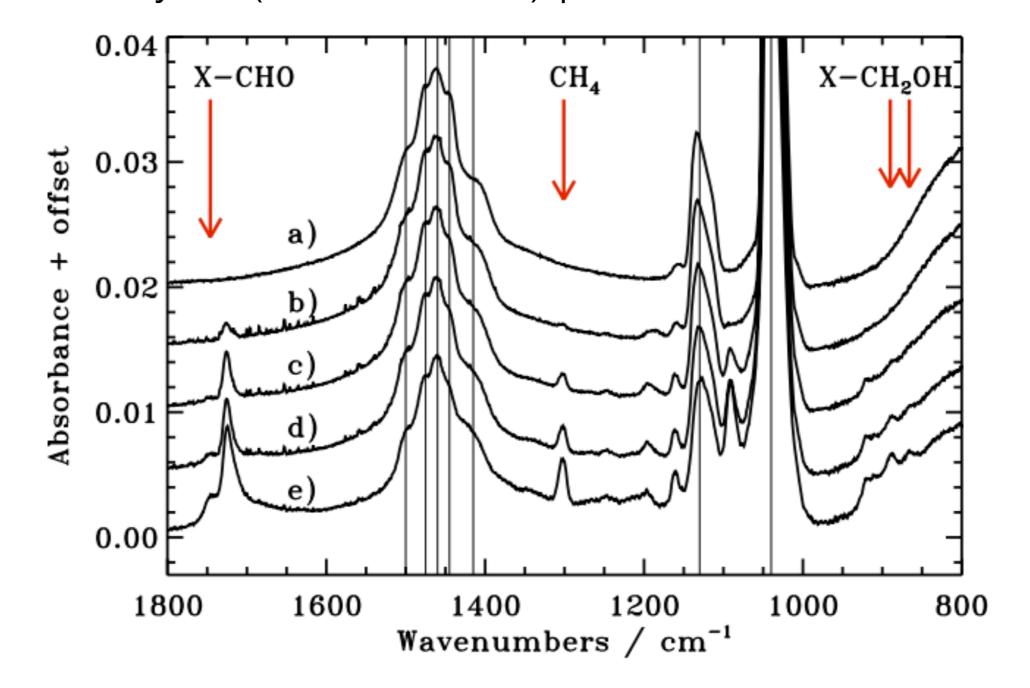


Formation H₂CO and CH₃OH: hydrogenation CO ice

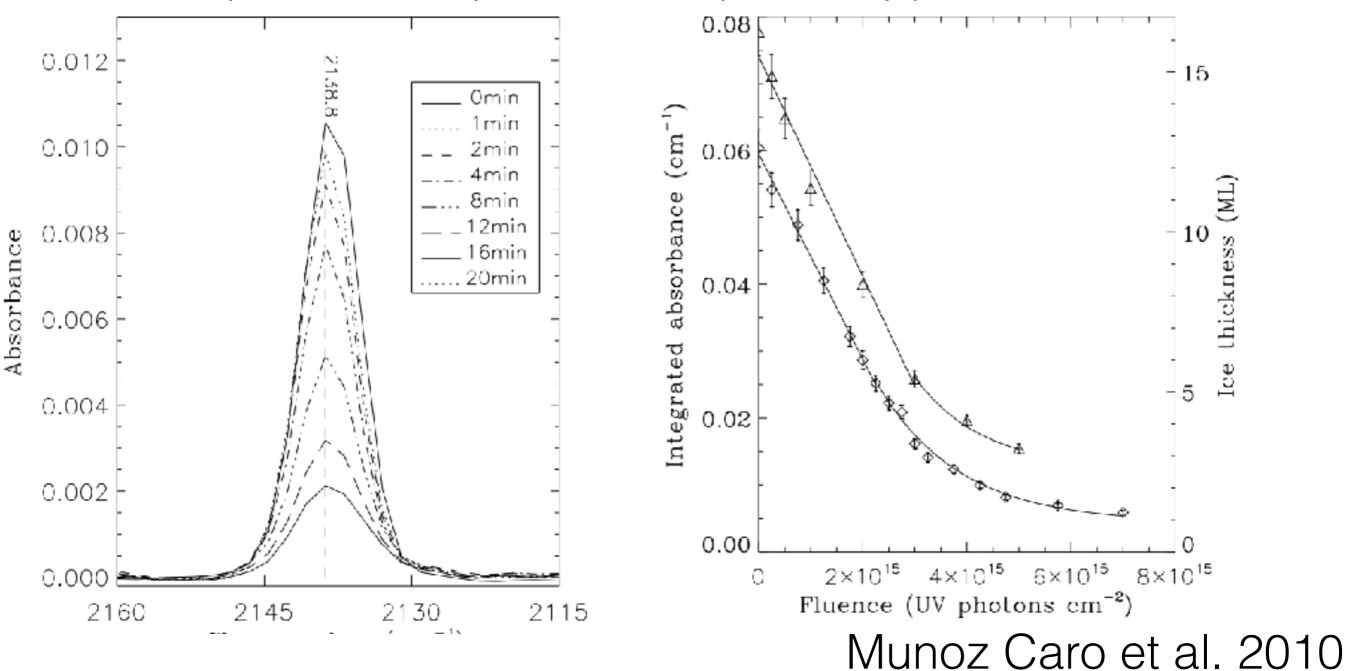


MC simulation

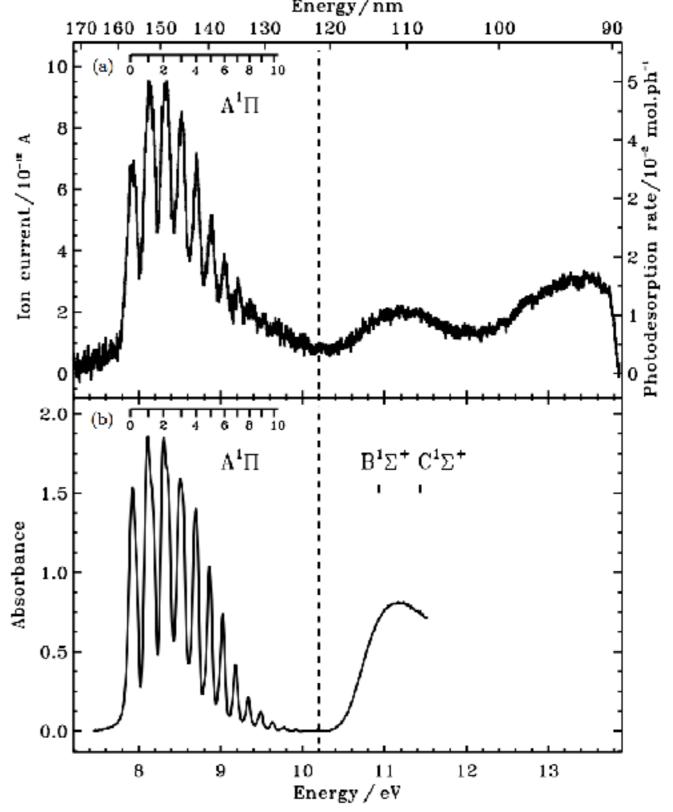
Photolysis (UV radiation) products of CH₃OH ice



CO photodesorption rates (UV lamp)

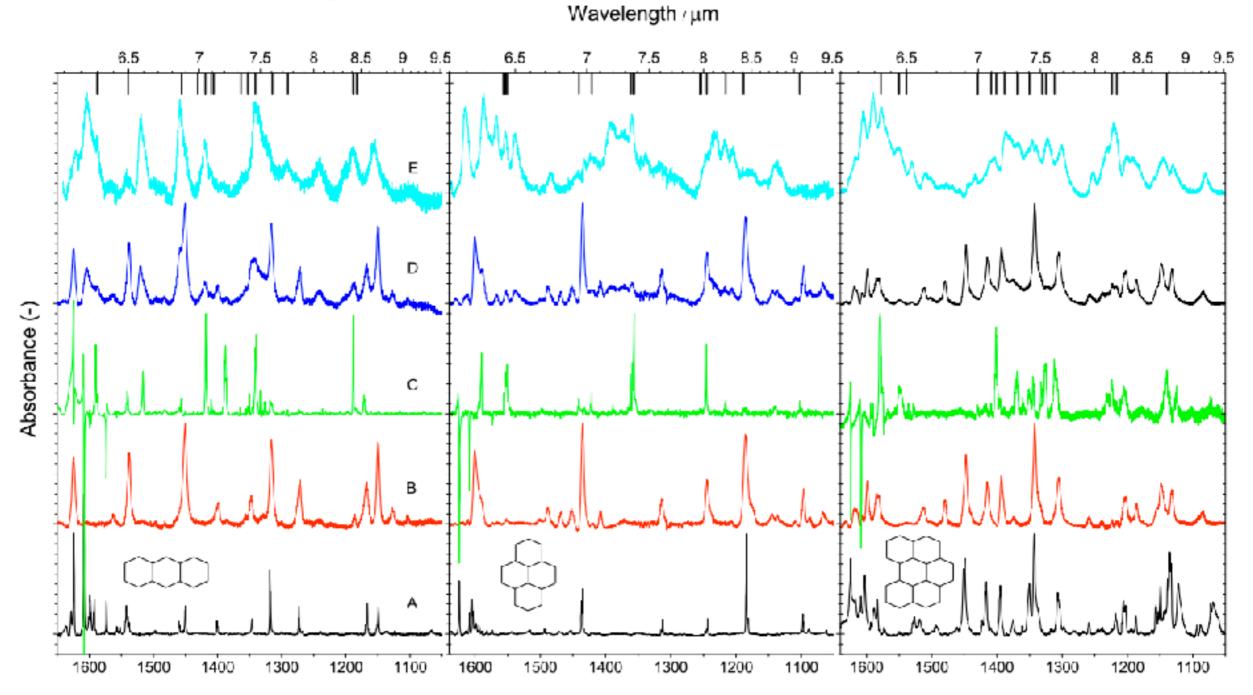


- UV-dependence studies: FUV beam line at SOLEIL synchrotron (Paris)
- Note wavelength range: UV (90-170 nm):
 - => electronic states

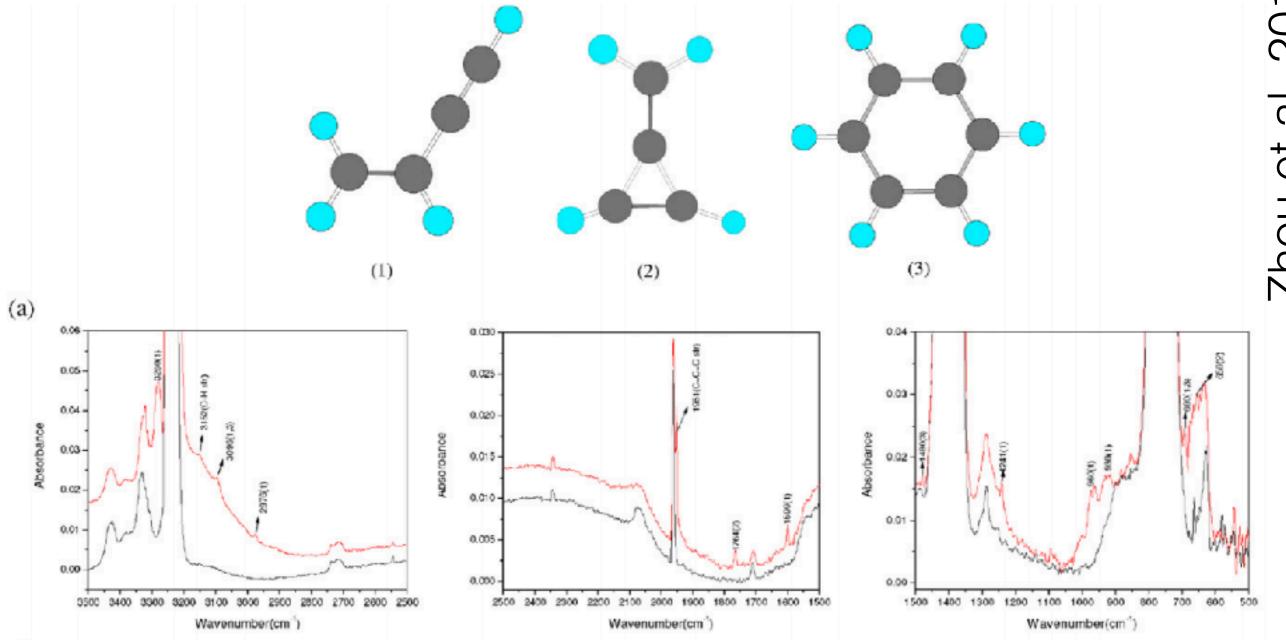


Fayolle et al. 2011

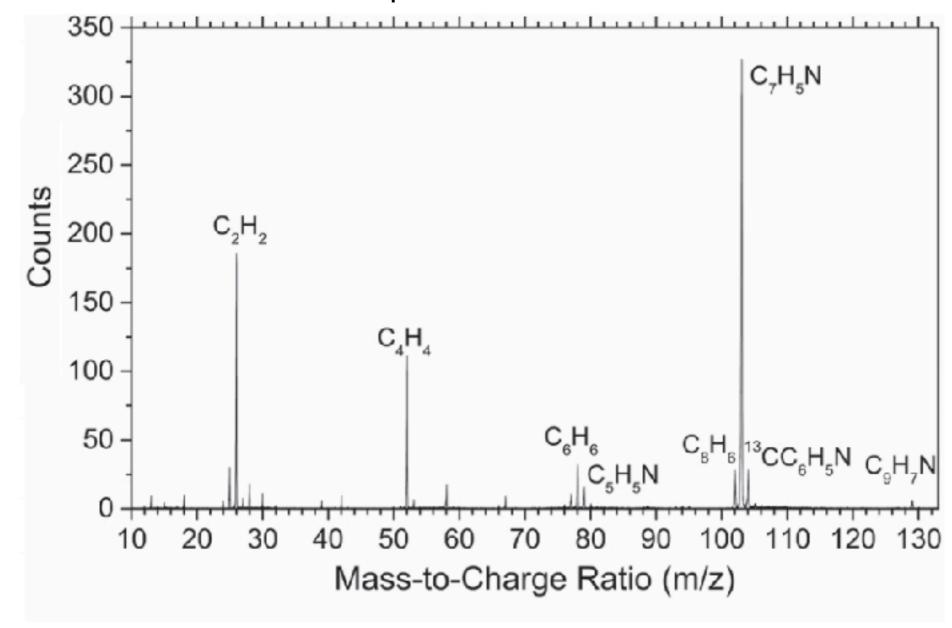
PAHs in H₂O matrices



Cosmic ray-induced formation of benzene (Titan)



 Nitrogen-PAHs formation around environments similar to envelopes of carbon stars



Astrochemistry labs

- Generally a mix of physicists, chemists and astronomers: very interdisciplinary! Also collaboration with modellers
- Experimental work: failures, calibration and money
- Examples labs:
 - Kaiser lab UH (Kaiser)
 - Strathclyde, Heriot-Watt in UK Virginia (Herbst)
 - Alabama (Gerakines)
 - NASA Ames (Sandford)
 - Harvard (Oberg)
 - Sackler lab Leiden (Linnartz)
 - Strathclyde, Heriot-Watt (McCoustra, Fraser)
 - Laboratoire d'Astrophysique Bordeaux, France (Wakelam)

http://www.astrochymist.org/astrochymist_whom.html

Talk Adi

Next week

- Using ALMA to look for molecules
 - Splatalogue
 - Basic interferometry
 - ALMA dataset
 - Imaging
 - Tutorial (please install CASA!)